UV spectroscopy and acid hydrolysis it may be concluded that the rhamnose in substance (I) is attached at position 3, and the only possible site of attachment of the glucose is position 8. On the basis of the results obtained we propose for substance (I) the structure of 7-methylgossypetin $8-\beta$ -D-glucopyranoside $3-O-\alpha$ -L-rhamnopyranoside, and for (II) 7-methylgossypetin $8-\beta$ -D-glucopyranoside.

V. I. Sheichenko and L. P. Smirnova took part in the recording of the NMR spectra of the substances obtained.

LITERATURE CITED

- 1. N. P. Maksyutina, Khim. Prirodn. Soedin., 62 (1965).
- 2. T. K. Chumbalov, M. M. Mukhamed'yarova, L. P. Smirnova, I. S. Chanysheva, and V. B. Omurkamzinova, Khim. Prirodn. Soedin., 658 (1976) [in this issue].

C-GLYCOSIDES OF Ajania fastigiata

T. K. Chumbalov and R. A. Zhubaeva

UDC 547.972

A methanolic extract from the epigeal mass of Ajania fastigiata family Compositae, was concentrated, and the chlorophyll was precipitated with water. The aqueous methanolic solution was distributed in organic solvents. From an ethereal extract, in addition to quercetin and luteolin [1], by preparative chromatography on paper we isolated a substance (I) with the composition $C_{15}H_{10}O_5$, mp 343-345°C (aqueous methanol), which proved to be apigenin, as was confirmed by the melting point of the acetyl derivative, the products of alkaline fusion, and the results of IR and UV spectroscopy.

The residual aqueous methanolic solution was chromatographed on Kapron. Elution with 20% methanol gave the total C-diglycosides [substances (II) and (II)], which were purified on a column of cellulose. Similar glycosides have previously been separated by preparative chromatography on paper [2]. We propose the use of Sephadex LH-20, which considerably shortens the time of separation and gives substances of higher purity.

The Sephadex was swollen in water, and water was also used for dissolving the substances and for elution from the columns.

Substances (II) and (III) had mp 228-230°C and 236-238°C (aqueous ethanol). The action of 5% HCl led to their mutual isomerization with the appearance of two new isomers, which is characteristic for C-diglycosides [3].

Compounds (II) and (III) did not undergo enzymatic hydrolysis [4]. On acid hydrolysis by Kiliani's method, apigenin, D-glucose, and traces of D-arabinose were detected [5].

IR spectrum of the C-glycosides, cm^{-1} : 3300-3400, 1650, 1620, 1570, 1520, 1450, 1075, 1045, 1020, 910.

UV spectrum [λ_{max} (absolute ethanol)] of substance (II): 332 and 280 nm (log ϵ 3.93; 3.89); substance III: 336 and 276 nm (log ϵ 3.91; 3.88). The ratios of the intensities of the absorption maxima in the long-wave region of the spectra of (II) and (III) were 35 and 32% of the intensity of the absorption maximum of the aglycone [6]. A reduced bathchromic shift with zirconyl chloride was observed: $\Delta\lambda + 23$ nm (II) and +20 nm (III) [7]; [α] $_{D}^{22} + 55^{\circ}$ (II) and $+99^{\circ}$ (III) (c 0.5%; dimethylformamide) [2]; [M] $_{D} \cdot K_{D} = +153.5$ (II) and +276.3 (III) [8].

Substance (II) was identified as apigenin 6,8-di-C- β -D-glucopyranoside and (III) as a rotational isomer of (II).

LITERATURE CITED

1. T. K. Chumbalov and R. A. Zhubaeva, Collection of Papers on Chemistry from Kazakh State University [in Russian, Alma-Ata, No. 3 (1973), p. 39.

S. M. Kirov Kazakh State University, Alma-Ata. Translated from Khimiya Prirodnykh Soedinenii, No. 5, pp. 661-662, September-October, 1976. Original article submitted August 25, 1975.

This material is protected by copyright registered in the name of Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$7.50.

- 2. T. K. Chumbalov and O. V. Fadeeva, Khim. Prirodn. Soedin., 364 (1970).
- 3. M. K. Seikel and T. J. Mabry, Tetrahedron Lett., 16, 1105 (1965).
- 4. J. B. Harborne, Phytochemistry, 4, 107 (1965).
- 5. V. I. Litvinenko and I. P. Kovalev, Khim. Prirodn. Soedin., 56 (1967).
- 6. V. I. Litvinenko, Rast. Res., 2, 4 (1964).
- 7. V. I. Litvinenko and V. M. Darmograi, Dokl. Akad. Nauk USSR, Ser. B., No. 7, 639 (1968).
- 8. I. P. Kovalev and V. I. Litvinenko, Khim. Prirodn. Soedin., 233 (1965).

A FLAVONOL GLYCOSIDE FROM PLANTS OF THE GENUS Phellodendron

V. É. Otryashenkova, V. I. Glyzin, and G. K. Shreter

UDC 547.972

A glycoside with the composition $C_{26}H_{32}O_{12}$, mp 205°C, λ_{max} 290, 345 nm has previously been isolated from Phellodendron amurense Rupr. (Amur cork tree) and has been named phellamurin. It has the structure of noricaritin 7-O- β -D-glycoside. A glycoside with the composition $C_{26}H_{32}O_{12}$, mp 154-156°C, has been isolated from P. japonicum Maxim (Japanese cork tree) and P. amurense, and this has been assigned the structure of β -amhydronoricaritin 3-O- β -D-glucoside and the name phellodendroside [2]. From the same plant has been isolated a glycoside with the composition $C_{32}H_{42}O_{17}$, mp 152-153°C, λ_{max} 225, 290, 345 nm having the structure of dihydronoricaritin 7, γ -di-O- β -glucopyranoside which has been called dihydrophelloside [3]. From P. lavallei Dode. (Lavalle cork tree) and P. amurense a glycoside has been isolated with the composition $C_{26}H_{32}O_{12}$, mp 151-153°C, λ_{max} 290, 345 nm, for which the structure of isonoricaritin 7-O- β -D-glucopyranoside and the name phellavin have been proposed [4].

Phellodendroside, dehydrophelloside, and phellavin have similar constants and are the main flavonol glycosides of the materials investigated. To compare their flavonol compositions, we studied six species of cork tree the constants of the substances obtained being given below. The initial material consisted of the leaves of the plants collected in Maritime Territory (P. amurense) and in the botanical garden of the Academy of Sciences of the Uzbek SSSR in Tashkent (the other plants):

Plant	Composition	mp, °C	$\lambda_{ ext{max}}$ (CH ₃ OH)
Phellodendron amur-	$C_{26}H_{32}O_{12}$	200-203	220, 291, 346
ense Rupr.	$C_{26}H_{32}O_{12}$	15 1- 154	220, 291, 346
P. japonicum			
Ma xi m	$C_{26}H_{32}O_{12}$	150-155	220, 291, 348
P. chinense Schneid.	$C_{26}H_{32}O_{12} \cdot 1H_2O$	150-153	220, 291, 346
P. Lavallei Dode.	$C_{26}H_{32}O_{12}$	150-152	220, 291, 346
P. sacchalinense	2		
(Fr. Schmidt) Sarg.	$C_{26}H_{32}O_{12} \cdot 1/2H_2O$	150-152	220, 291, 346
P. piriforme E. Wolf.	$C_{26}H_{32}O_{12} \cdot 1/2H_2O$	150-154	220, 291, 346

From the Amur cork tree together with the main glycoside (mp 151-154°C) we isolated a glycoside with the composition $C_{26}H_{32}O_{12}$, mp 200-203°C, which is probably phellamurin.

The glycosides from R. japonicum, P. chinense Schneid. (Chinese cork tree), P. sacchalinense (Fr. Schmidt.) Sarg. (Sakhalin cork tree), and P. piriforme E. Wolf (pearfruit cork tree) proved from their chromatographic mobilities [Silufol UV-254, chloroform—ethyl acetate—formic acid (6:6:3) system], melting points,

This material is protected by copyright registered in the name of Plenum Publishing Corporation, 227 West 17th Street, New York, N.Y. 10011. No part of this publication may be reproduced, stored in a retrieval system, or transmitted, in any form or by any means, electronic, mechanical, photocopying, microfilming, recording or otherwise, without written permission of the publisher. A copy of this article is available from the publisher for \$7.50.

I. M. Sechenov First Moscow Medical Institute. Translated from Khimiya Prirodnykh Soedinenii, No. 5, pp. 662-663, September-October, 1976. Original article submitted March 18, 1976.